

- 2 (a) Define "internal energy" and prove that it a property of a system. 6
- (b) A container is divided in to compartments by partition. 10
The container is completely insulated so that there is no heat transfer. One portion contains gas at temperature T_1 and pressure P_1 while the other portion also has the same gas but at the temperature T_2 and pressure P_2 .
If the partition is removed, what will be the change of internal energy and work done due to mixing of two gas?

OR

- 2 A cylinder contains $0.45m^3$ of gas at 1 bar and $80^\circ C$. The 16
gas is compressed to a volume of $0.13m^3$, the final pressure being 5 bar, determine
- (i) The mass of gas
(ii) the value of index 'n' for compression
(iii) The increase in internal energy of gas
(iv) The heat exchanged by the gas during compression.
- 3 (a) Write down the general energy equation for steady 9
flow system and simplify when applied to (i) centrifugal water pump (ii) steam nozzle.
- (b) In a gas turbine unit, the gas flows through the turbine 9
is 15 kg/s and the power developed by the turbine is 12 MW. The enthalpies of gas at inlet and outlet are 1260 kJ/kg and 400 kJ/kg respectively and the velocities of the gas at inlet and outlet are 50 m/s and 110 m/s respectively. Calculate the rate at which heat is rejected from the turbine.

- 4 Attempt the following : 18
- (a) Define partial pressure, partial volume, Dalton's law and Amagat's law.
 - (b) Differentiate between dry air and moist air ? Can we treat water vapour in moist air as perfect gas ? Justify your answer.
 - (c) Define following :
COP, Heat pump, Kelvin Plank statement, Clausius statement.
 - (d) A domestic food freezer maintains a temperature of -15°C . The ambient temperature is 30°C . The heat leaks into the freezer at 1.75 kJ/sec . What is the minimum power necessary to pump out heat.

- 5 (a) State and prove Clausius inequality. 16
- (b) A reversible heat engine operates between two reservoirs at 600°C and 40°C . The engine drives a reversible refrigerator which operates between the same 40°C reservoir and a reservoir at -18°C . The heat transfer to the heat engine is 2100 kJ and there is a net output of 370 kJ from combined plant. Evaluate the heat transfer to the refrigerator and net heat transfer to the 40°C reservoir.

OR

- 5 (a) Explain concept of Entropy in detail. 16
- (b) 4 kg of water at 27°C is mixed with 1 kg of ice at 0°C . Assuming adiabatic mixing determine the temperature of the mixture of water and ice. Calculate the net change of entropy. Assume enthalpy of fusion of ice as 335 kJ/kg .

6 Attempt any two :

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- (a) Explain following terms :
fire point, pour point, flash point, cloud point.
- (b) How are fuels classified ? Compare chemical and physical properties of any two fuels.
- (c) State and explain thermodynamic temperature scale.
- (d) Prove that $\eta_{Carnot\ cycle} = \frac{T_1 - T_2}{T_1}$ where T_1 is source temperature and T_2 is sink temperature.
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